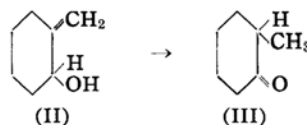


*Isomerization of Luciculine**

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Lucidusculine (I) ($C_{24}H_{35}O_4N$) is one of the aconite alkaloids; upon hydrolysis it gives a trihydric amino alcohol, luciculine (II) ($C_{22}H_{33}O_3N \cdot H_2O$)¹⁾. In contact with palladium-carbon in the presence of hydrogen and methanol, compound II was found to isomerize to a ketone base, isoluciculine (III), m. p. 131~136°C, λ_{max} in MeOH 302 m μ , $\log \epsilon$ 1.4 (*Anal.* Found: C, 71.22; H, 9.33; N, 3.32; H₂O, 2.33. Calcd. for $C_{22}H_{33}O_3N \cdot \frac{1}{2}H_2O$: C, 71.71; H, 9.30; N, 3.80; H₂O, 2.44%). Compound III showed a band at 5.82 μ in the infrared spectrum, and was considered therefore to be a cyclic six-membered ketone. Since it was shown previously that a terminal methylene group exists in compound II¹⁾, this isomerization seems to proceed similarly to that of Shimoburo base I²⁾, hypognavine³⁾ or napellonine^{4,**)} as follows:



On the basis of these observations, the presence of an allyl alcohol moiety in the structure of compound II may be claimed.

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** It has been shown that napellonine is identical with Shimoburo base I⁵⁾.

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